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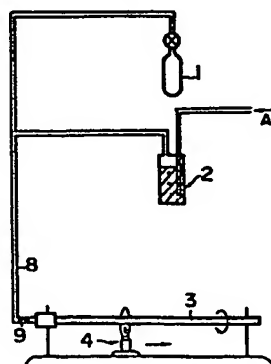
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(54) Infrared optical fiber.

(57) Solid or hollow infrared optical fiber of high stability and small light transmission loss is produced from halides and/or hydrides of elements constituting calcogenide glass as starting materials by MCVD process with less chance for contamination with impurities and with unnecessary for melting for a prolonged time for homogenization.

FIG. 1



INFRARED OPTICAL FIBER

1 BACKGROUND OF THE INVENTION

The present invention relates to a process for producing optical fiber capable of transmitting infrared beam.

5 Heretofore, optical fiber is produced from materials of silicate glass, but the materials of silicate glass have such a disadvantage that they transmit only light beams of up to wavelength of about 2 μm and absorb substantially all other beams of longer wavelength by absorption due to lattice vibration. Thus, beams of wavelength of 10.6 μm from a CO_2 laser for use, for example, in laser surgery and laser welding, etc. cannot be transmitted through the fiber of silicate glass. Accordingly, search has been so far made for 15 materials capable of transmitting beams of wavelength from about 2 μm to about 20 μm , and as a result calcogenide glass has been regarded as a promising material for the said range of wavelengths. For example, "Infrared Physics, 1965, Vol. 5 p.p. 69 - 80" discloses 20 infrared optical fiber of As-S glass. The conventional optical fiber of calcogenide glass is produced from powder of such metal as As, S, Ge, P, etc., or calcogen as a starting material by melting for a prolonged time. Thus, the conventional process has such a 25 disadvantage that impurities contained in the starting

- 1 material powder of metal or calcogen are transferred
into product glass, thereby increasing light trans-
mission loss. Furthermore, the conventional process
has more chance of contamination by impurities during
5 the melting, and also requires substantial time for
melting and homogenizing the powder of metal or calcogen.

The optical fiber is in such a structure
that the core is made from a material of higher refractive
index than that for the clad surrounding the core to
10 utilize total reflection at the interface between
the core and the clad to transmit beams. However,
when a beam of a high output level, for example, 100 W,
such as CO₂ laser beam is transmitted therethrough,
the core material itself reaches such a high temperature
15 as 100° - 400°C due to the absorption of beams by the
core material. Thus, there appear such phenomena that
the core material undergoes thermal breakage or deterio-
ration of mechanical strength. Furthermore, considerable
heat generation takes place particularly at the end
20 face of optical fiber, for example, due to fine dirt
thereon or due to fine irregularity thereon, and special
care is taken for its protection. Consequently, there
has been a restriction to the power level of input
beams.

- 25 On the other hand, MCVD process is well
known as one of the conventional processes for producing
optical fiber of silicate glass. According to the
MCVD process, a glass film serving as a clad and a

1 core, or a glass film serving as a core is deposited
onto an inner wall of a glass tube of quartz, etc. by
the so-called CVD process utilizing chemical vapor
reaction, then the glass tube is heated, thereby
5 collapsing the hollow part and forming a rod-like
optical fiber preform, and the preform is drawn with
heating, thereby forming optical fiber, as disclosed
in so many publications, for example, US Patent No.
4,203,743; Japanese Laid-open Patent Application No.
10 29953/76, etc.

The following references are cited to show
the state of art; i) Infrared Physics, 1964, Vol. 4
p.p. 213 - 221, ii) Infrared physics, 1965, Vol. 5,
p.p. 69 - 80, iii) US Patent No. 4,203,743, iv) Japanese
15 Laid-open Patent Application No. 29953/76, and v)
The Bell System Technical Journal, 1964, Vol. 43,
p.p. 1783 - 1809.

SUMMARY OF THE INVENTION

20 An object of the present invention is to
provide a process for producing infrared optical
fiber with low transmission loss, where the said
problems in producing infrared optical fiber from
calcogenide glass, that is, the problems of contami-
25 nation of impurities and melting for a prolonged time
have been solved.

Another object of the present invention
is to provide a process for readily producing infrared

- 1 optical fiber where the said problem of heat generation has been solved.

The foregoing objects can be attained according to the present process for producing infrared optical

5 fiber, which comprises i) a step of introducing into a glass tube a gas containing a halide and/or a hydride of elements constituting calcogenide glass as starting material, that is, a halide and a hydride of metal, particularly at least one element selected from the

10 group consisting of Si, Ge, As, P and Sb, and a halide or a hydride of at least one calcogen element selected from the group consisting of S, Se and Te, and heating the glass tube, thereby depositing the calcogenide glass, that is, a compound of said metal, etc. and said

15 calcogen element, onto the inside wall of the glass tube, and ii) a step of heating and drawing the resulting glass tube into fiber. In the step ii), the glass tube is heated at a high temperature to collapse the tube, and then drawn at a much higher temperature,

20 whereby solid optical fiber can be obtained. Hollow optical fiber can be also obtained by heating and drawing the glass tube in a hollow state without collapsing, and dry air or infrared transmission liquid can be filled into the hollow fiber. In the

25 case of solid fiber, the glass tube can be heated and drawn as such.

According to the present process, infrared optical fiber of high performance can be produced

1 without the problems, that is, without any fear of
contamination of impurities and any necessity for melting
for a prolonged time.

It is essential in the present process to
5 use a halide or a hydride of the metal, etc. or calcogen
element as a starting material in place of the powder
of the metal, etc. or calcogen element as in the
conventional process.

Furthermore, according to the present process,
10 glass breakage due to heat generation, which has been
a bottleneck in the prior art, can be avoided by using
hollow fiber, and infrared optical fiber capable of
transmitting a high energy level beam can be produced
thereby. This is based on transmission of HE_{11} mode
15 in a hollow waveguide of hollow optical fiber.

Optical fiber of high infrared transmittance
can be obtained by making the clad from calcognide
glass of high refractive index and high infrared
transmittance while using air as the core. A large
20 number of pseudo-transmitted modes (leaky modes)
can be transmitted through such hollow fiber. Different
from the conventional optical fiber with core of high
refractive index, the pseudo-transmitted modes are
waves transmitted while producing transmission loss,
25 among which the lowest order mode (HE_{11} mode) has the
least transmission loss.

According to "The Bell System Technical
Journal, 1964, Vol. 43, p.p. 1783 - 1807", the

- 1 transmission loss α of HE_{11} mode is given by the following formula:

$$\alpha = \left(\frac{2.405}{2\pi} \right)^2 \frac{\lambda^2}{a^3} \cdot \frac{\frac{1}{2}(n^2+1)}{\sqrt{n^2-1}} \dots (1)$$

5

where n represents a refractive index of glass, λ wavelength of light and a a radius of hollow part.

When $n = \sqrt{3}$ in formula (1), α takes the minimum value, i.e. $\alpha = 1.8 \lambda^2/a^3$ (dB). The refractive index of

- 10 calcogenide glass is usually $n = 25$, where the transmission loss is $2.01 \lambda^2/a^3$ (dB). When the wavelength of light is $10.6 \mu\text{m}$ and $a = 2 \text{ mm}$, the transmission loss will be 0.028 dB/m , which is small enough, as compared with the ordinary glass and crystal, and such
- 15 optical fiber can be used satisfactorily as a transmission path for high energy level beams.

- Another feature of the hollow fiber produced according to the present process is use of calcogenide glass of low transmission loss as a clad. Substantially
- 20 all of the energy of the beams transmitted through the fiber passes through the hollow part, and thus the electromagnetic field passing through the clad is small even with a high incident power, and even if the clad has transmission loss of 1 dB/m , the temperature
- 25 increase of the glass by absorption is small.

The said step i) can follow the well known MCVD process except for the starting material. When solid fiber is produced, the said step ii) can also

- 1 follow the well known MCVD process. When hollow optical fiber is produced, the glass tube from the step i) is drawn at such a heating temperature and in such a drawing ratio not to cause collapsing of
- 5 the glass tube while keeping the glass tube in a hollow state. The necessary heating temperature and the drawing ratio can be determined by simple preliminary test.

- Any material can be used for the glass tube,
- 10 so long as it can be drawn at a high temperature together with the deposited calcogenide glass. Usually, quartz glass, Pyrex glass, lead glass, etc. are used. When solid fiber is produced, collapsed optical fiber preform with the said glass tube as a crust can be drawn as
 - 15 such, but when the crust is removed from the preform by guiding or chemical etching and then the preform freed from the crust is drawn, infrared optical fiber consisting only of calcogenide glass can be obtained. In the latter case, any material can be used for the
 - 20 glass tube, so long as it can be collapsed at a high temperature together with the deposited calcogenide glass.

- When the starting material is in a gaseous state at room temperature in the said step i), it
- 25 can be directly introduced into the glass tube, but when it is in a liquid state at room temperature, it is bubbled with an inert gas such as Ar, etc., and the generated vapor is introduced into the glass tube

- 1 together with the inert gas as a carrier gas.

BRIEF DESCRIPTION OF THE DRAWINGS

5 Fig. 1 is a schematic diagram showing an apparatus for depositing calcogenide glass onto the inside wall of a glass tube by local heating of the glass tube according to one embodiment of the present invention.

10 Fig. 2 is a schematic diagram showing an apparatus for depositing calcogenide glass onto the inside wall of a glass tube by uniform heating of the glass tube.

15 Fig. 3 is a schematic diagram showing an apparatus for depositing calcogenide glass onto the inside wall of a glass tube whose heating part for reaction is separated from a deposition part.

Fig. 4 is a cross-sectional view of hollow fiber having a calcogenide glass layer on the inside wall of a glass tube.

20 Fig. 5 is a schematic view showing an apparatus for depositing calcogenide glass onto the inside wall of a glass tube by local heating of the glass tube according to another embodiment of the present invention.

25 DETAILED DESCRIPTION OF THE INVENTION

The present invention will be described in detail below by way of a case of producing As-S glass fiber, referring to the drawings.

- 1 In the production of As-S fiber glass,
AsCl₃ and H₂S are used as starting materials. AsCl₃
is in a liquid state at room temperature, and thus
a bubbler 2 is used as shown in Fig. 1, AsCl₃ is placed
5 in the bubbler 2 and bubbled with Ar. AsCl₃ is intro-
duced into a glass tube 3 through a gas line 8 together
with Ar. On the other hand, H₂S is in a gaseous state
at room temperature, and is led to the glass tube 3
through the gas line 8 from a cylinder 1 directly.
10 That is, a raw material gas of AsCl₃, H₂S and Ar in
mixture is led to the glass tube 3 through the gas
line 8 in the arrow direction 9.

Then, the raw material gas is heated in the
glass tube 3 and thermally decomposed to deposit As-S
15 onto the inside wall of the glass tube 3. Heating can
be carried out by an electric furnace or a gas burner.
In Fig. 2 heating is carried out by a gas burner 4.

In Fig. 1, the glass tube 3 is locally heated
by the gas burner 4, and the gas burner 4 is moved
20 in the longitudinal direction of the tube at the same
time to obtain a deposit film of uniform composition
and uniform thickness in the longitudinal direction.
It is also possible to rotate the tube to form a uniform
deposit in the circumferential direction.

25 Such local heating is however one example,
and the glass tube 3 can be heated to the full length
of glass tube by an electric furnace 5, as shown in
Fig. 2, or a heating part 6 for chemical vapor reaction

1 of the raw material gas can be separated from a deposi-
tion part 7 usually at a lower temperature than that
of the heating part, as shown in Fig. 3. Anyway, the
raw material gas mixture is introduced into the glass
5 tube 3 through the gas line 8 in the arrow direction
9.

The glass tube onto whose inside wall As-S
glass is deposited in the foregoing manner is heated
and drawn into fiber. In the production of solid fiber,
10 usually the As-S glass-deposited glass tube is heated
and collapsed to form optical fiber preform, and then
the preform is drawn at a much higher temperature,
whereas in the production of hollow fiber, the As-S
glass-deposited glass tube is directly heated and drawn
15 so that no collapsing can occur. In the production of
solid fiber, the As-S glass-deposited glass tube can
be also directly heated and drawn without forming the
preform.

First of all, the production of solid fiber
20 will be described in detail below:

The As-S glass-deposited glass tube is
further heated for collapsing. Heating for the collapsing
can be local heating by a burner, where the heating
part is moved in the longitudinal direction of the glass
25 tube. In that case, heating can be uniformly carried
out in the circumferential direction of the glass
tube by rotating the glass tube. Collapsing can
proceed under the inside pressure of the glass tube

- 1 equal or a little inferior to the atmosphere, but the
As-S glass is liable to vaporize, the tube inside is
made vacuum and the ends of the tube are sealed before
the collapsing is carried out. The thus prepared
5 solid rod is called "preform".

Then, the preform is drawn into fiber in
the same manner as in the conventional drawing of a
preform of quartz glass fiber. In that case, the
resulting fiber has oxide glass, which is the constituent
10 of the glass tube 3, as a crust. By removing the oxide
glass from the preform by guiding or chemical etching
beforehand, optical fiber consisting only of As-S
glass can be obtained.

The foregoing process is directed to the
15 production of As-S glass fiber having a uniform com-
position in the radial direction. Clad-type fiber
of double layer structure consisting of a core of high
refractive index and a clad of low refractive index,
or graded index-type fiber whose refractive index is
20 gradually increased from the peripheral part toward
the center can be produced by changing the composition
in the radial direction in the following manner.

The composition can be changed in the radial
direction by reciprocating the burner of heating the
25 glass tube while moving in the longitudinal direction
of the glass tube several times in the longitudinal
direction, while changing the composition of raw
material gas to the glass tube at every reciprocation

1 or at a specific reciprocation. The composition of
raw material gas can be changed by changing a mixing
ratio of AsCl_3 to H_2S , or by adding other additive
such as GeCl_4 thereto to change the concentration.

5 Production of hollow fiber will be described
in detail below.

The As-S glass-deposited glass tube is drawn
in the same manner as in the conventional drawing of
quartz glass fiber. In that case, heating temperature
10 and drawing ratio must be so selected as not to collapse
the glass tube. The thus produced optical fiber is
hollow fiber having As-S glass on the inside wall of
the glass fiber, as shown in Fig. 4, where numeral 10
is an air core, numeral 11 a calcogenide layer, and
15 numeral 12 a drawn glass tube.

In the foregoing, only the case of using
 AsCl_3 and H_2S as the starting materials has been
described for both solid fiber and hollow fiber.
Besides the foregoing starting materials, AsH_3 , AsBr_3 ,
20 AsI_3 , etc. can be also used for As, and S_2Cl_2 , SCl_2 , etc.
for S.

In the foregoing, description has been made
only of the fiber of As-S glass. The present process
is however not limited only to the As-S glass fiber,
25 but also to other calcogenide glass fiber. Other
calcogenide glass than the As-S glass includes: (1)
 Ge-Sb-S ; (2) Ge-Sb-Se ; (3) Ge-As-S ; (4) Ge-As-Se ;
(5) Ge-S-P ; (6) Ge-Se-P , etc. Typical composition

- 1 of calcogenide glass for use in the present process is combinations of at least one of Si, Ge, As, P and Sb with at least one of S, Se, and Te. Besides the foregoing elements, Cl, Br or I can be added thereto.
- 5 The present process is applicable not only to the compositions as defined above, but also to calcogenide glass of other compositions as defined above.

As described above, the present process has such advantages as less chances for contamination

10 of a product with impurities and unnecessary for melting for a prolonged period for homogenizing the composition, and is very effective for production of optical fiber with low transmission loss.

15 DESCRIPTION OF THE PREFERRED EMBODIMENTS

Example 1

- GeCl₄ and SbCl₅ were used as compounds of metal, etc, and Se₂Cl₂ as a calcogen compound as starting materials, and the apparatus shown in Fig. 5 was used.
- 20 GeCl₄ was placed in a bubbler 2 and bubbled with Ar. SbCl₅ was placed in a bubbler 2' and bubbled with Ar. Se₂Cl₂ was placed in a bubbler 2" and bubbled with Ar. A raw material gas of GeCl₄, SbCl₅ and Se₂Cl₂ carried by carrier Ar gas was introduced into a glass
- 25 tube 3 of lead glass (outer diameter: 14 mm, inner diameter: 12 mm) through a gas line 8 in the arrow direction 9. GeCl₄, SbCl₅ and Se₂Cl₂ were all in a liquid state at room temperature and were bubbled at

- 1 20°C. The composition of the raw material gas was controlled by bubbling of Ar to 43 mg/s of GeCl_4 , 30 mg/s of SbCl_5 and 137 mg/s of Se_2Cl_2 .

Chemical vapor reaction was carried out by using an oxygen-hydrogen burner 4 reciprocating at a rate of 0.2 cm/s in the longitudinal direction of the glass tube 3, while rotating the glass tube 3 at 40 rpm. Heating temperature was found to be 600°C by an optical pyrometer. After the heating oxygen-
10 hydrogen burner 4 was reciprocated in the longitudinal direction of the glass tube 3 by 50 repetitions, the glass tube was fused at about 800°C to collapse the glass tube into a preform.

- Then, the thus obtained preform was drawn
15 at 800°C into optical fiber with a Ge-Sb-Se glass part having an outer diameter of 50 μm and a crust glass part having an outer diameter of 150 μm .

The glass tube consisted of lead glass, and the calcogenide glass consisted of 28 mol.%Ge -
20 12 mol.%Sb - 60 mol.%Se. The thus obtained optical fiber had light transmission loss of 0.1 dB/m at the wavelength of 10.6 μm , and thus the infrared optical fiber with considerably smaller light transmission loss than that of the conventional optical
25 fiber was obtained.

Example 2

A Ge-Sb-Se glass film was deposited on the

- 1 inside wall of a glass tube in the same manner as in
Example 1, except that the glass tube of lead glass
having an outer diameter of 14 mm and an inner diameter
of 13 mm was used and the heating oxygen-hydrogen
5 burner 4 was reciprocated in the longitudinal direction
of the glass tube by 100 repetitions.

The calcogenide glass film-deposited glass
tube was heated in an electric furnace at 800°C and
drawn into hollow glass fiber. The hollow glass fiber
10 had an outer diameter of 4 mm, an inner diameter of
1.5 mm and a calcogenide glass film thickness of
300 μm.

The lead glass of the glass tube consisted
of 57 mol.%SiO₂ - 12 mol.%Na₂O + K₂O - 30 mol.%PbO,
15 and the deposited calcogenide glass had the same
composition as in Example 1.

The thus obtained hollow infrared optical
fiber had light transmission loss of 0.7 dB/m at the
wavelength of 10.6 μm, and had a temperature elevation
20 by only a few tens of degree °C even with an incidental
CO₂ laser beam of 50 W, and thus the infrared
optical fiber with a high stability and small light
transmission loss, as compared with those of the
conventional optical fiber was obtained.

25

Example 3

In the foregoing Example 1, solid optical
fiber having a uniform composition in the radial

1 direction was produced, whereas in the present Example
3, glass fiber having changing composition in the
radial direction was produced by using the same apparatus
as shown in Fig. 5 and used in Example 1.

5 In the present Example 3, 43 mg/s of GeCl_4 ,
60 mg/s of SbCl_5 and 137 mg/s of Se_2Cl_2 as the starting
materials were used up to 25th repetition of reciprocal
movement of the burner in the longitudinal direction
of the glass tube, and 43 mg/s of GeCl_4 , 30 mg/s of
10 SbCl_5 and 137 mg/s of Se_2Cl_2 were used from 26th
repetition to 50th repetition. Other conditions were
the same as in Example 1.

The thus obtained preform was drawn into
fiber.

15 An optical fiber having light transmission
loss of 0.01 dB/m at the wavelength of 10.6 μm was
obtained.

It seems that considerably smaller light
transmission loss in Example 3 than in Example 1 is
20 due to the fact that the present fiber consisted of
a core and a clad and had no substantial influence
by the surrounding glass tube.

The thus obtained clad-type fiber had a core
diameter of 50 μm , a clad thickness of 50 μm and total
25 diameter of 150 μm , and the core consisted of 28
mol.%Ge - 12 mol.%Sb - 60 mol.%Se, and the clad
consisted of 23.5 mol.%Ge - 12 mol.%Sb - 64.5 mol.%Se.

Obviously many modifications and variations

- 1 of the present invention are possible in the light of
the above teachings. It is therefore to be understood
that within the scope of the appended claims the
invention may be practised otherwise than as specifically
5 described.

CLAIMS:

1. A process for producing infrared optical fiber which comprises:

i) a step of introducing a gas containing halides and/or hydrides of elements constituting calcogenide glass into a glass tube and heating the glass tube, thereby depositing onto the inside wall of the glass tube calcogenide glass resulting from chemical vapor reaction, and

ii) a step of heating the glass tube from the step i) and drawing the tube into fiber at a high temperature.

2. The process according to Claim 1, wherein the calcogenide glass is a compound of at least one element selected from the group consisting of Si, Ge, As, P and Sb, and at least one element selected from the group consisting of S, Se and Te.

3. The process according to Claim 1 or 2, wherein the step ii) comprises heating the glass tube, thereby forming a solid optical fiber preform, and drawing the preform into solid fiber at a high temperature.

4. The process according to Claim 1 or 2, wherein the step ii) comprises drawing the glass tube into hollow fiber without collapsing the glass tube.

5. The process according to Claim 3, wherein the preform is drawn into fiber at a high temperature after the crust of the preform has been removed.

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FIG. 1

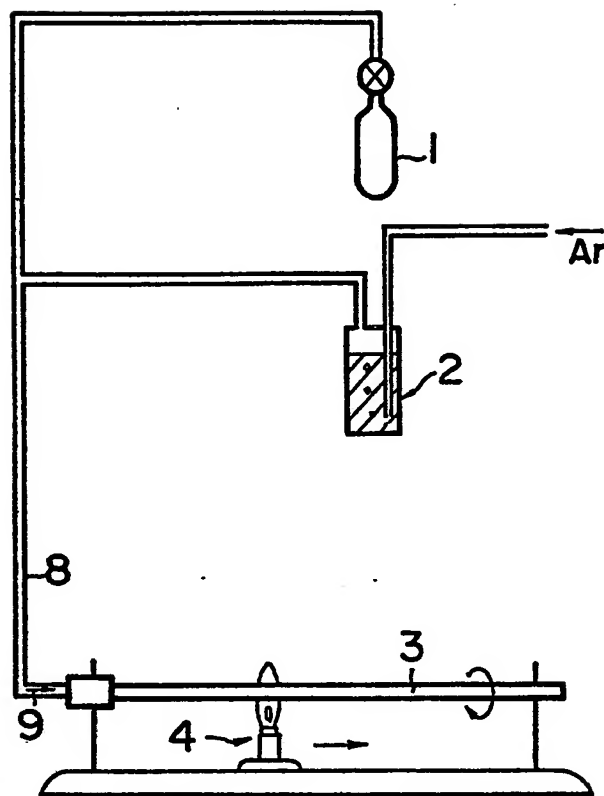


FIG. 2

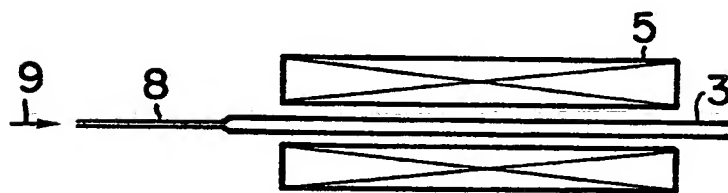
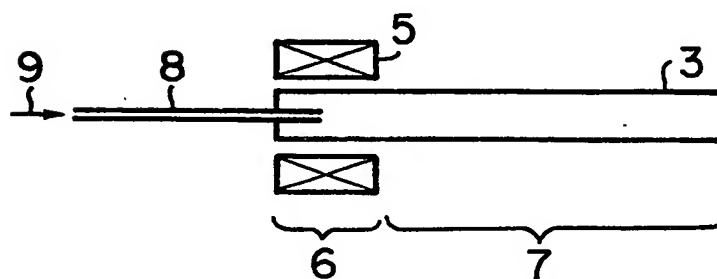


FIG. 3



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FIG. 4

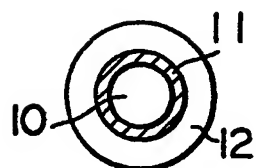
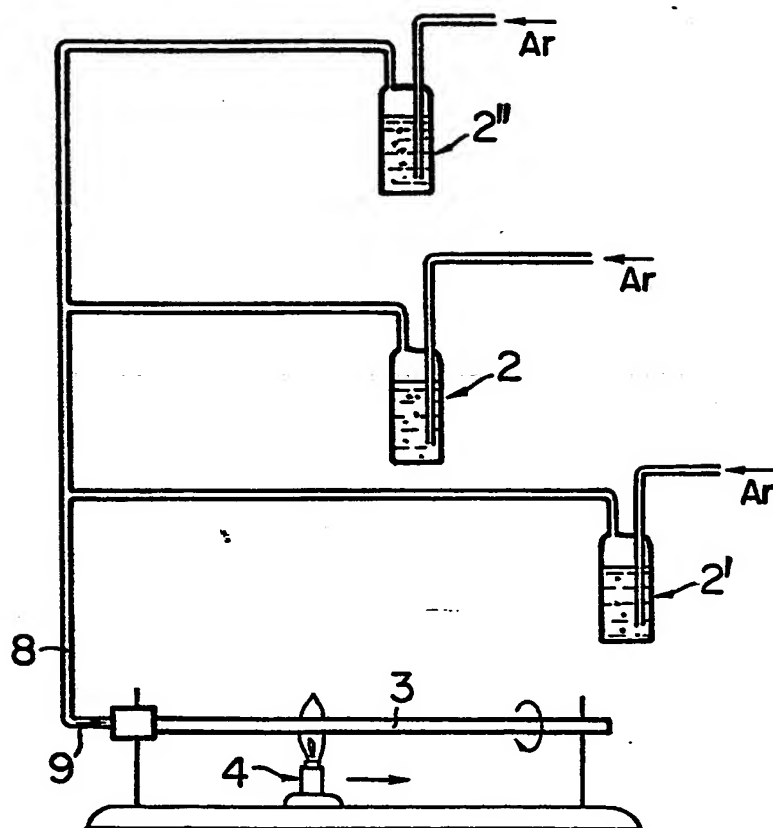


FIG. 5





European Patent
Office

EUROPEAN SEARCH REPORT

0060085

Application number

EP 82 30 1088

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 7)
X	US-A-4 188 089 (G. GLIEMER ROTH) * claims 1,2,4,6,7,12 *	1-3	G 02 B 5/172 C 03 B 37/02
A	US-A-3 980 459 (TINGYE LI) * summary; figures *	1	
A	FR-A-2 384 722 (I.S.E.C.) * claims *	1	
A	DE-B-2 546 162 (JENAER GLASWERK) * columns 3,4 *	1	
A,D	US-A-4 203 743 (T. SUGANUMA et al.) * abstract *	1	
			TECHNICAL FIELDS SEARCHED (Int. Cl. 7)
			G 02 B 5/172 C 03 B 37/02
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 17-06-1982	Examiner PEAHLER R.
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	